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S-14 1. Solubility of Ethanol in WaterExplain why ethanol (CH 3CH 2OH) is more soluble in water than is ethane (CH 3CH 3). Answer Ethanol is polar; ethane is not. The ethanol —OH group can hydrogen-bond with water. 2. Calculation of pH from Hydrogen Ion ConcentrationWhat is the pH of a solution that has an H concentration of (a) 1.75 10⁻⁵ mol/L; (b) 6.50 10⁻¹⁰ mol/L; (c) 1.0 10⁻⁴ mol/L;

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