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Modern Chemistry Chapter 7 test. STUDY. PLAY. The basis of an ionic bond is the__ absorption of water into their solid structures. Elements tend to gain or lose electrons in order to acquire the electron configuration of a__ noble gas. Ions that are formed from more than one atom are called__

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chemical formula subscript of each element reveals number of atoms contained in a single molecule of a certain compound monoatomic ions ions formed from a single atom; remember to change nonmetals to -ide ending oxyanions polyatomic ions that contain oxygen binary ionic compounds compounds of 2 elements between metal (positive, s-block and p-block) and nonmetal [...]

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Modern Chemistry Chapter 7. monatomic ion. binary compounds. nomenclature. oxyanions. an ion formed from a single atom. (remember to change nonmetal.... compounds composed of two different elements; (compounds of 2.... naming system. polyatomic ions that contain oxygen.

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CHAPTER 7 REVIEW Chemical Formulas and Chemical Compounds MIXED REVIEW SHORT ANSWER Answer the following questions in the space provided. 1. Write formulas for the following compounds: CuCO₃ a. copper(II) carbonate Na₂SO₃ b. sodium sulfite (NH₄)₃PO₄ c. ammonium phosphate SnS₂ d. tin(IV) sulfide HNO₂ e. nitrous acid 2.

7 Chemical Formulas and Chemical Compounds

CHAPTER 7 REVIEW Chemical Formulas and Chemical Compounds SECTION 1 SHORT ANSWER
Answer the following questions in the space provided. 1. c In a Stock system name such as iron(III) sulfate, the Roman numeral tells us (a) how many atoms of Fe are in one formula unit. (b) how many sulfate ions can be attached to the iron atom. (c) the charge on each Fe ion.

7 Chemical Formulas and Chemical Compounds

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Modern Chemistry 35 Chapter Test Name Class Date Chapter Test B, continued Write the orbital notation for the following elements in the space provided Modern chemistry chapter 7 test answer key. 37. lithium, atomic number 3 38. carbon, atomic number 6 39 Modern chemistry chapter 7 test answer key. neon, atomic number 10 PART VI Write the answers to the questions on the line to the left, and ...

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Modern Chemistry 141 Chapter Test Name Class Date Chapter Test B, continued 18. When a strong acid is titrated with a weak base, the pH of the solution at the equivalence point is than 7. 19. A is a highly purified solid used to check the concentration of a standard solution. 20. A 1 M solution of NaOH will have a pH that is

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Modern Chemistry • CHAPTER 7 HOMEWORK 7-6 SKILL BUILDER Determine the oxidation numbers of these compounds and ions. N₂ N₃-N₂O₃ KNO₃ NH₄NO₃ K₂Cr₂O₇ CO₃²⁻ NH₄⁺ ... Circle the letter of the best answer. 1. Which is the Stock system name of SO₂? a. sulfur dioxide b. sulfur trioxide c. sulfur(IV) oxide d. sulfur(II) oxide

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