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SECTION 3 SHORT

ANSWER Answer the following questions in the space provided. 1. c When an electron is added to a neutral atom, energy is (a)

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always absorbed. (c) either absorbed or released. (b) always released. (d) neither absorbed nor released.

2. d The energy required to remove an electron from a neutral atom is the atom's

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ANSWER Answer the following questions in the space provided. 1.

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ANSWER Answer the following questions in the space provided. 1.

c In the modern periodic table, elements are ordered (a) according to decreasing atomic mass. (b) according to Mendeleev's original design.

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ANSWERAnswer the following questions in the space provided. 1.

When an electron is added to a neutral atom, energy is .

(a)always absorbed(c)either absorbed or released.

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(b)always
released(d)burned
away. 2.

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SHORT ANSWER

Answer the following questions in the space provided. 1. Consider the neutral atom with 53 protons and 74 neutrons to answer the following questions. 53.

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a. What is its atomic number? 127. b. What is its mass number? atomic number. c. Is the element's position in a modern periodic table ...

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chxistina. Terms in this set (36) The order of elements in the periodic table is based on atomic mass the number of protons in the nucleus the number of neutrons in the nucleus

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statement is true about the metal... At room temperature none of the metals.... Which general statement does NOT apply.... a weighted average of the masses of an element's isotopes. Silicon is a better conductor of electric current than sulfur....

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c When an electron is added to a neutral atom, energy is . (a) always absorbed (c) either absorbed or released (b) always released (d) burned away 2. d The energy required to remove an electron from an atom is the atom's .

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Answers 5 1

Chapter 5 Review The Periodic Law Answers 5 1 the modern periodic law states that the physical and chemical properties of an element are functions of its atomic number the discovery of the noble gases changed

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SECTION 5-3 SHORT

ANSWER Answer the following questions in the space provided 1

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