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Phase Change Worksheet Answers

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Before you answer any further questions, please turn on the heat and watch the molecules and the graph. Think about the definitions of the different phases of matter (solid, ... PHASE CHANGE: Simulation worksheet 2. What was difficult about the procedure (measuring the temperature, keeping the time, ...

PHASE CHANGE: Simulation worksheet

Phase Change Worksheet The graph was drawn from data collected as a substance was heated at a constant rate. Use the graph to answer the following questions. 1 80 Phæe Change 60 1 40 1 20 100 60 40 12 Tim e (minutes) 20 22 At point A, the beginning of observations, the substance exists in a solid state. Material in this phase has volume and shape.

Nicolet High School

Heat with Phase Change Worksheet 1) How many joules are required to heat 250 grams of liquid water from 00to 1000C ? 2) How many joules are required to melt 100 grams of water? 3) How many joules are required to boil 150 grams of water?

Heat with Phase Change Worksheet

The phase change from water vapor to liquid water is known as... Preview this quiz on Quizizz. The phase change from water vapor to liquid water is known as... Phase Changes DRAFT. ... answer choices . evaporation. precipitation. condensation. sublimation. Tags: Question 2 . SURVEY . 60 seconds . Q.

Phase Changes | States of Matter Quiz - Quizizz

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KEY PHASE DIAGRAM WORKSHEET Part A - Generic Phase Diagram. Answer the questions below in relation to the following generic phase diagram. 1. Which section represents the solid phase? A 2. What section represents the liquid phase? C 3. What section represents the gas phase? B 4.

Phase Diagram Worksheet - Studylib

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Calculating Energy for Changes of Phase To Calculate Heat: Heat to Change the Temperature: $Q = m \cdot T \cdot s$ Heat to Melt: $Q = m \cdot H_{fus}$ Heat to Boil: $Q = m \cdot H_{vap}$ For Water: $H_{fus} = 334 \text{ J/g}$ $H_{vap} = 2260 \text{ J/g}$ Specific Heat: ice = $2.06 \text{ J/g } ^\circ\text{C}$ water = $4.184 \text{ J/g } ^\circ\text{C}$ steam = $2.03 \text{ J/g } ^\circ\text{C}$

Calculating Energy for Changes of Phase

(answer in terms of potential and/or kinetic energy) ____ The energy is causing the position of the atoms to change (P.E. is increasing) there is a phase change ____ c) What phase change, if any, is taking place?

Heating Curve Worksheet - Studylib

Question: Worksheet Directios Fer H Hange Only, With No Phase Change Then Name: Per: USE Q-m CAT Change Involved, Temp. Remains Constant

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Then USE Q-m AHas For Melting (E Absorbed) Or Treezing CE . If Phase Involved, Released) If Temp, Crosses F Or USE Q-m AHap For Boiling (E Absorbed) Or Condensing (E Released) Ranges Related Hom Gas To Liquid To Solid Then Must ...

Worksheet Directios Fer H Hange Only, With No Phas ...

A student worksheet (Phase Change Extend) can be found on Page 8. Note: If students have not yet earned Achievements, remind them that they need to select the Achievement first, in order to receive credit for completion.

Collisions Lesson Plan Energy & Phase Change

Changes of State Worksheet Write the correct phase change based on the diagram at the top of the page. (snowman, pot of boiling water, floating in the water) 1. Sublimation 2. Melting 3. Evaporation 4. Solidification 5. Condensation 6. Deposition 7. Complete the table, using the examples inserted below. Change of state Heat added or released?

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